

A NEW SYNTHETIC METHOD FOR  $\beta$ -TRIMETHYLSILYL- $\alpha,\beta$ -UNSATURATED CARBONYL  
 COMPOUNDS EMPLOYING 1-METHOXY-3-PHENYLTHIO-3-TRIMETHYLSILYL-1-PROPENE

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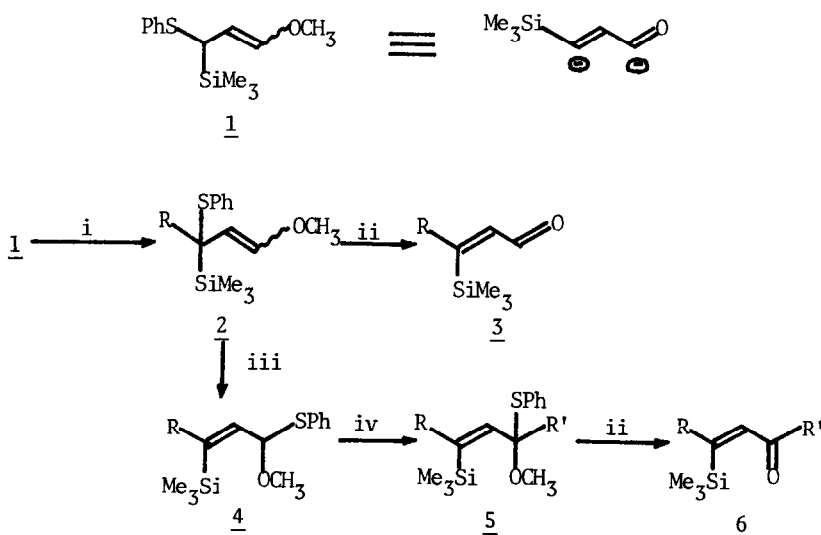
Summary: A new synthetic method for functionally substituted  $\beta$ -trimethylsilyl- $\alpha,\beta$ -unsaturated carbonyl compounds from 1-methoxy-3-phenylthio-3-trimethylsilyl-1-propene is described.

$\beta$ -Trimethylsilyl- $\alpha,\beta$ -unsaturated carbonyl compounds have proven to be attractive because of both their unique spectral properties and synthetic utilities, though there have appeared only a few general synthetic methods for them.<sup>1)</sup>

Due to their synthetic utilities, a variety of ambident allylic anions have received much attention.<sup>2)</sup> Among them, of special interest is an ambident allylic anion stabilized with both sulfur and oxygen because of its possibility of incorporation of electrophiles in a completely regioselective manner.<sup>3)</sup>

We wish to report here that the allylic sulfide 1<sup>4)</sup> is a new homoenolate dianion equivalent<sup>3c)</sup> with a trimethylsilyl group which would make a various type of  $\beta$ -trimethylsilyl- $\alpha,\beta$ -unsaturated carbonyl compounds available. Our synthetic course is shown in Scheme.

Scheme




i) n-BuLi, TMEDA, HMPA, RX, THF ii) NaIO<sub>4</sub>, dioxane-water iii) SiO<sub>2</sub>-hexane  
 iv) n-BuLi, HMPA, R'X, THF

Alkylation of 1 results in an exclusive formation of  $\alpha$ -adduct 2 that can be converted into 3 on treatment with  $\text{NaIO}_4$  via a [2,3]-sigmatropic rearrangement. On the other hand, a novel silica gel-promoted thioallylic rearrangement of 2 followed by  $\alpha$ -alkylation of resultant 4 enables incorporation of two electrophiles in the desired position in 1. Another notable feature of the present method consists in a mild oxidation process, thus functionally substituted products 3 or 6 being successfully obtained.

Reaction of the carbanion of 1 ( $n\text{-BuLi}$ , TMEDA,  $-20^\circ\text{C}$ , 0.5 h) with various alkyl halides in the presence of HMPA took place smoothly at  $-78^\circ\text{C}$  for 0.5 h and at  $-40^\circ\text{C}$  for 1 h, providing 2 in good yields. Sequential desulfurization of 2 thus obtained occurred easily through oxidation with  $\text{NaIO}_4$  (1.2 eq) in dioxane-water (5:1) at room temperature for 1 h to afford 3. On the other hand, the thioallylic rearrangement of 2 took place with ease on heating a hexane solution of 2 (1.0 mmol) under reflux for 4 h in the presence of silica gel (flame dried in vacuo, 2.0 g). The newly formed allylic sulfide 4, pure enough without purification, was treated with  $n\text{-BuLi}$  (1.2 eq) in THF at  $-40^\circ\text{C}$  for 1 h followed by HMPA (1.2 eq) and alkyl halides (1.2 eq) at  $-40^\circ\text{C}$  for 1 h to afford 5. Crude 5 thus obtained was treated with  $\text{NaIO}_4$  in the same manner mentioned above to give 6 in good yields. These results are summarized in Table.

Table Conversion of 1 into 3 or 6

entry	RX	R'X	Yield (%) <sup>a</sup>			
			<u>2</u>	<u>3</u>	<u>4</u>	<u>6</u> <sup>b</sup>
1	$\text{CH}_2=\text{CHCH}_2\text{Br}$		69 <sup>c</sup>	62		
2	$\text{THPO}(\text{CH}_2)_3\text{Br}$	$n\text{-C}_6\text{H}_{13}\text{Br}$	79	72	89	62
3	$\text{CH}_2=\text{CH}(\text{CH}_2)_9\text{I}$	$\text{THPO}(\text{CH}_2)_3\text{Br}$	75	80	88	56
4	$n\text{-C}_6\text{H}_{13}\text{Br}$	$\text{THPO}(\text{CH}_2)_3\text{Br}$	80 <sup>d</sup>	75	92	55
5	$n\text{-C}_6\text{H}_{13}\text{Br}$	 $\text{CH}(\text{CH}_2)_2\text{Cl}$	80 <sup>d</sup>	75	83	48

a) Isolated yield after column chromatography (silica gel). b) Yield based on 4. c) Isolated yield after distillation (Kugelrohr temp.  $100^\circ\text{C}/0.6\text{ mmHg}$ ). d) Isolated yield after distillation (Kugelrohr temp.  $130^\circ\text{C}/0.6\text{ mmHg}$ ).

#### References

1) J. Otera, T. Mandai, M. Shiba, T. Saito, K. Shimohata, K. Takemori, and Y. Kawasaki, *Organometallics*, **2**, 332 (1983), and references cited therein. 2) For the most recent paper on this subject; Y. Yamamoto, H. Yatagai, Y. Saito, and K. Maruyama, *J. Org. Chem.*, **49**, 1096 (1984). 3)(a) M. Wada, H. Nakamura, T. Taguchi, and H. Takei, *Chem. Lett.*, 345 (1977); (b) T. Mandai, M. Takeshita, M. Kawada, and J. Otera, *Chem. Lett.*, 1259 (1984); (c) T. Mandai, T. Moriyama, Y. Nakayama, K. Sugino, M. Kawada, and J. Otera, *Tetrahedron Lett.*, **25**, 5913 (1984). 4) The allylic sulfide 1 was prepared as follows: Addition of  $\text{Me}_3\text{SiCl}$  (3.0 eq) in one portion to a THF solution of the carbanion of 3-methoxy-1-phenylthio-1-propene<sup>3a)</sup> (LICA,  $-78^\circ\text{C}$ , 0.5 h) and HMPA (1.2 eq) followed by stirring at  $-78^\circ\text{C}$  for 1 h gave 1 in 85% yield after distillation (Kugelrohr temp.  $85^\circ\text{C}/0.6\text{ mmHg}$ ).

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